



Analytical methods

Quantification of N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL) and Total Lysine through Stable Isotope Dilution Assay and Tandem Mass Spectrometry

Antonio Dario Troise, Alberto Fiore, Markus Wiltafsky, Vincenzo Fogliano

PII: S0308-8146(15)00707-4

DOI: <http://dx.doi.org/10.1016/j.foodchem.2015.04.137>

Reference: FOCH 17538

To appear in: *Food Chemistry*

Received Date: 14 January 2015

Revised Date: 28 April 2015

Accepted Date: 29 April 2015

Please cite this article as: Troise, A.D., Fiore, A., Wiltafsky, M., Fogliano, V., Quantification of N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL) and Total Lysine through Stable Isotope Dilution Assay and Tandem Mass Spectrometry, *Food Chemistry* (2015), doi: <http://dx.doi.org/10.1016/j.foodchem.2015.04.137>

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

Quantification of Nε-(2-Furoylmethyl)-L-lysine (furosine), Nε-(Carboxymethyl)-L-lysine (CML), Nε-(Carboxyethyl)-L-lysine (CEL) and Total Lysine through Stable Isotope Dilution Assay and Tandem Mass Spectrometry

Antonio Dario Troise^{1,2*}, Alberto Fiore³, Markus Wiltafsky⁴, Vincenzo Fogliano¹

¹ Food Quality Design Group, Wageningen University, PO Box 8129, 6700 EV, Wageningen, The Netherlands

² Department of Agricultural and Food Science, University of Napoli "Federico II", Parco Gussone, 80055 Portici, Napoli, Italy

³ School of Science, Engineering & Technology Division of Food Science, Abertay University Dundee DD1 1HG, UK

⁴ Evonik Industries AG, Rodenbacher Chaussee 4, 63457 Hanau, Germany

***Corresponding author**

Antonio Dario Troise

antonio.troise@wur.nl

Phone number: +39 081 2539360

Abstract

The control of Maillard reaction (MR) is a key point to ensure processed foods quality. Due to the presence of a primary amino group on its side chain, lysine is particularly prone to chemical modifications with the formation of Amadori products (AP), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL). A new analytical strategy was proposed which allowed to simultaneously quantify lysine, CML, CEL and the N ϵ -(2-Furoylmethyl)-L-lysine (furosine), the indirect marker of AP. The procedure is based on stable isotope dilution assay followed by, liquid chromatography tandem mass spectrometry. It showed high sensitivity and good reproducibility and repeatability in different foods. The limit of detection and the RSD% were lower than 5 ppb and below 8%, respectively. Results obtained with the new procedure not only improved the knowledge about the reliability of thermal treatment markers, but also defined new insights in the relationship between Maillard reaction products and their precursors.

Keywords: Maillard reaction, LC-MS/MS, CML, CEL, lysine, furosine

Abbreviations: Maillard reaction (MR), Maillard reaction end products (MRPs) N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL),

List of compounds: N ϵ -(2-Furoylmethyl)-L-lysine (furosine, PubChem CID: no items) , N ϵ -(Carboxymethyl)-L-lysine (CML, PubChem CID: 123800), N ϵ -(Carboxyethyl)-L-lysine (CEL, PubChem CID: no items), lysine (PubChem CID: 5962).

1. Introduction

The final quality of many industrial food products depends on food formulation and processing design resulting in the formation of a huge variety of molecules as a consequence of thermal treatments and chemical changes (van Boekel, Fogliano, Pellegrini, Stanton, Scholz, Lalljie, et al., 2010). Along with lipid oxidation, the Maillard reaction (MR) occupies a prominent place in the final quality of food being responsible not only for the desired color and aroma compounds but also for the formation of potentially toxic Maillard reaction end products (MRPs). The reaction between reducing sugars and amino groups is the first step in the Maillard cascade: the formation of the stable 1-amino-1-deoxy-2-ketose the Amadori product (AP) and 2-amino-2-deoxyaldose Heyns products represents the starting point of the many chemical pathways of this reaction (Hodge, 1953). The presence of an amino group on the side chain of lysine makes this amino acid particularly sensitive to the carbonyls attachments. The modifications arising from the lysine blockage resulted in the formation of a bewildering array of molecules: N ϵ -(1-Deoxy-D-fructos-1-yl)-L-lysine (fructosyl-lysine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL), pentosidine, pyrrolidine, lysino-alanine, 5-hydroxymethylfurfural (HMF), α -dicarbonyls and aroma key odorants (Yaylayan & Huyghuesdespointes, 1994). Fructosyl-lysine, CML and CEL represent the most widely studied MRPs, and they are often used as biomarker of food quality (Erbersdobler & Somoza, 2007; Nguyen, van der Fels-Klerx, & van Boekel, 2013). As highlighted in **Figure 1**, the acid hydrolysis adopted to release free amino acids from the polypeptide chain promote the conversion of the 1-deoxy-fructosyl-L-Lysine (AP) through a cyclized Schiff base, into the of N ϵ -(2-furoylmethyl)-L-lysine (furosine) which is a compound that can be quantified after protein hydrolysis and it has been widely used as marker of thermal treatment particularly in the dairy products (Krause, Knoll, & Henle, 2003).

The formation of CML and CEL from the oxidation of ARP and HRP has been well characterized (Nguyen, van der Fels-Klerx, & van Boekel, 2013). Carbohydrate fragmentation allows the

formation of glyoxal and methylglyoxal that readily react with lysine residues yielding the glycoxidation products CML and CEL, respectively (Ahmed, Thorpe, & Baynes, 1986). Moreover, CML and CEL can be formed via the Namiki-pathway through three subsequent steps: Schiff base production, glycolaldehyde alkylimine synthesis, oxidation and formation of glyoxal or methylglyoxal which react with lysine to yield CML and CEL. Another route of CML and CEL formation is linked to lipid peroxidation as glyoxal and methylglyoxal can derive from polyunsaturated fatty acids (Hidalgo & Zamora, 2005). Moreover, the two markers can be also formed from fragmentation and subsequent glycation of ascorbic acid and dehydroascorbic acid (Leclerc, Birlouez-Aragon, & Meli, 2002).

From the analytical point of view the identification of these markers of heat treatment can be approached in several ways (Tessier & Birlouez-Aragon, 2012). Furosine is used as indirect marker of quality control of moderately heat-treated dairy samples. The golden standards for furosine detection are ion-exchange chromatography, reverse phase high performance liquid chromatography (RP-HPLC) with UV detection, (Henle, Zehetner, & Klostermeyer, 1995) capillary electrophoresis and ion-pairing HPLC by using sodium-heptanosulphonate (Vallejo-Cordoba, Mazorra-Manzano, & Gonzalez-Cordova, 2004). These procedures had several drawbacks mainly related to the modifications occurring during sample preparation: the acidic hydrolysis does not allow the differentiation between AP and glycosyl-amine; overestimation or underestimation linked to the acidic hydrolysis might occur due to the formation of further intermediates and end-products (Pischetsrieder & Henle, 2012).

As for furosine, CML and CEL analysis implies acidic hydrolysis to hydrolyze peptide bonds followed by their quantification that could be performed by different instrumental methods (Nguyen, van der Fels-Klerx, & van Boekel, 2013). In some papers a pre-column derivatization with o-phthalaldehyde was used to allow the detection by fluorescence detector (Hartkopf, Pahlke, Ludemann, & Erbersdobler, 1994), while a widely used approach for CML and CEL detection is

gas or liquid chromatography coupled with tandem mass spectrometry. Specifically, multiple reaction monitoring (MRM) mode improves the sensitivity, reduces the coefficient of variability and ruled out the problems of derivatization (Delatour, Hegele, Parisod, Richoz, Maurer, Steven, et al., 2009). A double derivatization is required for GC separation and this bottleneck highlights the advantages of LC-MS/MS detection: no derivatization, highest sensitivity and good reproducibility (Charissou, Ait-Ameur, & Birlouez-Aragon, 2007; Fenaille, Parisod, Visani, Populaire, Tabet, & Guy, 2006). Moreover CML, CEL and lysine detection is possible also by matrix-assisted laser desorption/ionization time-of-flight mass spectrometry (MALDI-TOF-MS) that allows relative quantification of protein lactosylation and it is a reliable method to monitor the early Maillard reaction as well as MRPs during milk processing (Meltretter, Becker, & Pischetsrieder, 2008).

The aim of the present paper, was to further improve the existing methodologies for the detection of lysine and MRPs. A new method was designed which included direct hydrolysis along with stable isotope dilution assay coupled with solid phase extraction and ion pairing liquid chromatography tandem mass spectrometry (LC-MS/MS). The developed procedure allowed the simultaneous detection of total lysine, furosine, CML and CEL. The method was tested on several foods: milk, infant formulas, cookies, bread slices. The robustness after several injections and the reliability of the results obtained were evaluated in soybean-based feed products obtained under severe thermal treatment conditions. Data demonstrated satisfactory analytical performances on all tested samples and results were perfectly in line with those previously obtained.

2. Material and methods

2.1 Chemicals and reagents

Acetonitrile, methanol and water for solid phase extraction (SPE) and LC-MS/MS determination were obtained from Merck (Darmstadt, Germany). The ion pairing agent perfluoropentanoic acid, trichloroacetic acid, hydrochloric acid (37%) and the analytical standards L-lysine hydrochloride

and [4,4,5,5- d_4]-L-lysine hydrochloride (d_4 -Lys) were purchased from Sigma-Aldrich (St. Louis, MO). Analytical standards N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine (CML) and its respective deuterated standard N ϵ -(Carboxy[$^2\text{H}_2$]methyl)-L-Lysine (d_2 -CML) were obtained from Polypeptide laboratories (Strasbourg, France), N ϵ -(Carboxyethyl)-L-lysine and its internal standard N ϵ -(Carboxy[$^2\text{H}_4$]ethyl)-L-lysine (d_4 -CEL) were purchased from TRC-Chemicals (North York, Canada).

2.2 Foods samples

Powdered infant formula and milk samples were purchased in a local market, biscuits samples and bread slices were prepared according previous papers published by our group (Fiore, Troise, Mogol, Roullier, Gourdon, Jian, et al., 2012; Vitaglione, Lumaga, Stanzione, Scalfi, & Fogliano, 2009). UHT milk was prepared according to the procedure previously described (Troise, Fiore, Colantuono, Kokkinidou, Peterson, & Fogliano, 2014). Raw milk (protein, 3.5%; fat, 1%) was purchased in a local market.

2.2.1 Soybean samples

One batch of quartered raw soybeans was purchased from Rieder Asamhof GmbH & Co. KG (Kissing, Germany). The raw soybeans were further processed at the hydrothermal cooking plant of Amandus Kahl GmbH & Co. KG (Reinbeck, Germany). First, the beans were short-term conditioned to reach a temperature of 80 °C after 45 seconds. Afterwards, the beans entered a hydrothermic belt cooker at 72 °C and left inside for 3 min at a temperature of 70 °C. Then they were expanded at 117 °C using an annular gap expander (Typ OEE 8, Amandus Kahl GmbH & Co. KG, Reinbeck, Germany). The expanded soybeans were collected in a drying wagon for 10 min. Then they were dried with air at 65 °C for 10 min and cooled for another 10 min to reach a final moisture content of 12%. Afterwards, the expanded soybeans were autoclaved for 0, 5, 10, 15, 20,

25, 30, 35, 40, 45, 50 and 60 min at 110 °C and 1470 mbar using a fully controlled autoclave (Typ HST 6x9x12, Zirbus Technology GmbH, Bad Grund, Germany).

2.3 Samples preparation

Lysine and its derivatives N ϵ -(2-Furoylmethyl)-L-lysine (furosine), N ϵ -(Carboxymethyl)-L-lysine (CML), N ϵ -(Carboxyethyl)-L-lysine (CEL) were analyzed considering previous papers (Delaunay, et al., 2009; Fenaille, Parisod, Visani, Populaire, Tabet, & Guy, 2006; Troise, Dathan, Fiore, Roviello, Di Fiore, Caira, et al., 2014) and introducing several modifications. Briefly, 100 mg of each sample was accurately weighed in a screw capped flask with PTFE septa and 4 mL of hydrochloric acid (6 N) was added. The mixture was saturated by nitrogen (15 min at 2 bar) and hydrolyzed in an air forced circulating oven (Mettler, Schwabach, Germany) for 20 h at 110° C. The mixture was filtrated by polyvinylidene fluoride filters (PVDF, 0.22 Millipore, Billerica, MA) and 400 μ l was dried under nitrogen flow in order to prevent the oxidation of the constituents. The samples were reconstituted in 370 μ l of water and 10 μ L of each internal standard d_4 -Lys, d_2 -CML and d_4 -CEL was added in order to obtain a final concentration of 200 ng/mg of samples for both standards. Samples were loaded onto equilibrated Oasis HLB 1 cc cartridges (Waters, Wexford, Ireland) and eluted according to the method previously described, then 5 μ l was injected onto the LC/MS/MS system.

2.4 Liquid chromatography tandem mass spectrometry (LC-MS/MS)

Separation of furosine, CML, CEL, lysine and their respective internal standards was achieved on a reversed – phase core shell HPLC column (Kinetex C18 2.6 μ m, 2.1 mm x 100 mm, Phenomenex, Torrance) using the following mobile phases: A, 5 mM perfluoropentanoic acid and B, acetonitrile 5 mM perfluoropentanoic acid. The compounds were eluted at 200 μ L/min through the following gradient of solvent B (t in [min]/[%B]): (0/10), (2/10), (5/70), (7/70), (9/90), (10/90), (12/10), (15/10). Positive electrospray ionization was used for detection and the source parameters were

selected as follows: spray voltage: 5.0 kV; capillary temperature: 350 °C, dwell time 100 ms, cad gas and curtain gas were set to 45 and 5 (arbitrary units). The chromatographic profile was recorded in MRM mode and the characteristic transitions were monitored in order to improve selectivity using an API 3000 triple quadrupole (ABSciex, Carlsbad, CA). All relevant parameters are summarized in **Table 1**.

2.5 Analytical performances

CML, CEL, furosine and total lysine were quantified using a linear calibration curve built with specific solutions of CML spiked with d_2 -CML, lysine and furosine spiked with d_4 -lysine and CEL spiked with d_4 -CEL (final concentration of internal standards: 200 ng/ml) dissolved in water. The limit of detection (LOD) and the limit of quantitation (LOQ) were monitored according to the signal to noise ratio (Armbruster, Tillman, & Hubbs, 1994). The coefficients of determination r^2 for the 4 analytes were tested plotting the ratio between the pure compounds and their respective internal and the concentration of the pure compounds in the linearity range 5-1000 ng/mL. The internal standard ratio was used for the quantification and the relative standard deviation of intraday and interday assay was monitored three times each day and six times in different days. The recovery test was monitored according to the concentration of the internal standards used and to the ratio between labeled compounds and native compounds.

2.6 Statistical analysis

All of the analyses were performed in quadruplicate and the results expressed as mg/100 g of protein. Statistical calculations were performed using Matlab R2009b (Natick, MA) while for mass spectrometry data, Analyst version 1.4.2 (Applied Biosystems, Carlsbad, CA) was used.

3. Results and discussion

3.1 Liquid chromatography set up

Under the above described chromatographic conditions, typical retention time of CML and d_2 -CML was 7.11 min, for d_4 -Lys and Lys it was 7.23 min, for furosine it was 7.91 min, while for CEL and d_4 -CEL it was 7.36 min (**Figure 2**). Previous papers highlighted the problems due to the poor retention of amino acids and their derived molecules on silica bonded and C-18 column (Frolov & Hoffmann, 2008). Preliminary trials performed using C-18 column without the ion pairing agent confirmed this feature: the retention was poor and the analytes co-eluted with the impurities on the front of the chromatographic run with the consequent partial suppression of the signal associated to the markers. Inadequate separation of the analytes was obtained also using polar end-capped column; however a significant improvement was obtained using with this column perfluoropentanoic acid as ion pairing agent. In these experimental conditions, the retention time followed a typical reversed phase profile according to the polarity and to the steric hindrance of each molecule, as previously observed by other papers published earlier (Fenaille, Parisod, Visani, Populaire, Tabet, & Guy, 2006; Troise, Fiore, Roviello, Monti, & Fogliano, 2014). The presence of the ion pairing agent charged the core shell residues increasing the retention and promoting the selectivity of the positively charged CML, CEL, furosine, lysine and their respective internal standards. The presence of a core shell phase increased of the resolution which directly reflects the good performances of the reported method, the shape of the peak was maintained over each batch and the retention time shift was lower than 0.5 min, highlighting the robustness of the analytical performances.

3.2 Mass spectrometry set up

Mass spectrometry conditions were optimized by infusing singularly the seven standards directly in the ion source. Collision energy, declustering potential, tube lens voltage along with spray voltage and interface temperature were monitored in order to favor the formation of the typical fragmentation pattern (Delatour, et al., 2009). The lysine derived compounds underwent the formation of the fragment ion at 130 m/z which corresponds to the pipecolic acid generated by the

subsequent cyclization of the side chain of lysine and the loss of ϵ - amino group, similarly the mass shift for deuterated standards *d4*-CEL and *d4*-Lys was +4 Da as consequence of the fragmentation occurred on the side chain of lysine (**Figure S1** in supplementary material section) (Yalcin & Harrison, 1996). The MRM revealed the loss of formic acid giving the typical fragment at m/z 84; the mass shift for the deuterated molecules was +4. The seven standards were also infused inside the chromatographic flow in order to evaluate the interferences due to the ion pairing agent or to the solvent and the results revealed that no enhancement or suppression effect can be ascribed to the parameters monitored.

3.3 Analytical performances

The analytical performances of the method were tested against reproducibility, repeatability, limit of detection (LOD), limit of quantification (LOQ), linearity, precision, carry-over and coefficient of correlation (r^2). Before and after each batch, three solutions of acetonitrile and water (90:10; 50:50 and 10:90) were injected in order to verify the absence of any contaminants with the same signal and the same retention time of the analyzed molecules. The limit of detection and the limit of quantitation were determined according to the procedure previously described. The concentration 0.1 ppb resulted in no signal, while the LOD was 0.5 ppb for CML and lysine while for CEL and furosine it was 1 and 3 ppb, respectively. The slight differences among CML, CEL and furosine can be related to the different stability in the injection conditions. By injecting these concentrations the signal to noise ratio was always higher than 3. The LOQ were 5 ppb for CML, CEL and Lysine while for furosine it was 9 ppb, as highlighted in **Table 2**. These values were perfectly in line with those previously described for CML, CEL and lysine quantification by MS/MS (Delatour, et al., 2009; Tareke, Forslund, Lindh, Fahlgren, & Ostman, 2013) while for furosine the performance of LOD and LOQ were below the values previously reported in milk (Bignardi, Cavazza, & Corradini, 2012). According to the LOD and LOQ, linearity was achieved in the range 5-1000 ppb for CML, CEL and lysine, while for furosine the linearity range was between 9 and 1000 ppb. The carryover

effect was tested injecting after each point of the calibration curves a solution consisting in acetonitrile and water (50:50, v/v) and verifying the absence of the target compounds. The linearity of the calibration curves was evaluated three times in the same day (intraday assay for the reproducibility) and three times for three subsequent days (interday assay for the repeatability) using the ratio between the target compounds and their respective internal standard. The RSD (%) among the three curves was always lower than 8%, demonstrating that external factors had marginal impact on the performance of the method. Each point of the calibration curves was monitored using two specific transitions: the most intensive fragment was used as quantifier, the lowest as qualifier. For CML, CEL, furosine and lysine, the respective transitions of m/z 205–84.1, m/z 219.1–84.1, m/z 255.1–130.2, and m/z 147.2–130.2 were used as quantifier, whereas m/z 205–130.2, m/z 219.1–84.1, m/z 255.1–84, and m/z 147.2–84.1 were used as qualifier. CML was quantified using d_2 -CML as internal standard (m/z 207–144.1 and 207–84 for quantification and confirmation, respectively), CEL was quantified using d_4 -CEL (m/z 223–134.1 and 223 – 84 for quantification and confirmation, respectively) whereas for furosine and lysine, d_4 -lysine was used (m/z 151.2–134.1 and m/z 151.1– 88 for quantification and confirmation, respectively). The use of d_4 -lysine as internal standard for the quantification and recovery of furosine was optimized by monitoring the relative intensity of furosine standard towards d_4 -CEL, d_2 -CML and d_4 -lysine. A mixture of the four standards (10 ppm) was directly infused in the ion source. Results revealed that the intensity of the signal at m/z 151.2 and m/z 255 were similar and both were 15% higher than the signal of d_2 -CML and d_4 -CEL.

The response of the method in food was tested during each batch evaluating the ratio between the target compounds and the internal standard, these procedures confirmed and deepened the aspects linked to the recovery assay: in each sample the ratio between the area of the analyte and the area of the deuterated compounds was compared towards the calibration curve in order to obtain the final concentration of the analytes in the matrix. The intensity of the internal standard in the samples and

in the standard was compared and the RSD (%) between the spiked samples and the spiked standards was always lower than 10%. The recovery test was monitored in all the food matrix according to the intensity of the internal standard, the results were 91.1 ± 8.4 , 84.2 ± 7.4 , 88.0 ± 6.9 for *d2*-CML, *d4*-CEL and *d4*-Lysine.

3.4 CML, CEL, furosine and total Lysine in food

Powdered samples were freeze dried prior analysis in order to remove the interferences due to the humidity. The extraction procedure of MRPs is characterized by three key steps: the reduction with sodium borohydride, the hydrolysis with hydrochloric acid and the stable isotope dilution assay prior ion pairing solid phase extraction. According to the nature of protein and to their concentration each of the above listed can influence the yield and the efficiency of the extraction. The reduction with sodium borohydride promotes the conversion of free fructosyl-lysine into hexitol-lysine in order to avoid the overestimation of CML, CEL (Niquet-Leridon & Tessier, 2011). Moreover, the use of sodium borohydride is recommended when the concentration of free unstable Amadori products is high. Unfortunately, the use of this reducing agent had several drawbacks: protein degradation and free counterpart losses during the reduction, precipitation and purification procedure; moreover, the use of sodium borohydride can interfere with the release of furosine with the above mentioned reduction of fructosyl-lysine into hexitol-lysine. After several preliminary measurements it was decided to avoid the reduction. A good compromise between the detection of furosine and that of CML/CEL was achieved controlling the oxidation under nitrogen. In particular, prior the acidic hydrolysis the screw capped flasks were saturated with nitrogen in order to reduce the effect of autoxidation and control the reaction pathway (Yaylayan & Huyghuesdespointes, 1994).

The use of hydrochloric acid is a mandatory step for the hydrolysis of peptide bonds and for the release of amino acids, MRPs and for the conversion of fructosyl-lysine into furosine. Different

244 concentrations of protein per mL of hydrochloric acid can lead to different efficiency of the
 245 hydrolysis with the consequent underestimation of lysine content. In the present study, the
 246 extraction procedure was optimized in order to promote the dehydration reaction that leads to the
 247 formation of furosine and to the release of MRPs (Krause, Knoll, & Henle, 2003; Mossine &
 248 Mawhinney, 2007). Further studies will be conducted in order to compare the effect of time and
 249 concentration of hydrochloric acid on lysine release, mainly in protein rich samples.

250 The above described analytical performances were tested in food and feed samples in order to
 251 verify the robustness of the method. Several thermally treated foods were tested: powdered infant
 252 formula, low lactose milk, lab scale UHT milk, biscuits samples, bread (all prepared according to
 253 three different procedures previously described by our group) and powdered soybean-based feed
 254 products (prepared at industry scale). All data are summarized in **Table 3**. The concentration of
 255 CML in powdered infant formula analyzed ranged from 8.22 ± 0.31 mg/100 g of protein to $14.81 \pm$
 256 0.92 mg/100 g of protein, while CEL and furosine ranged from 0.71 ± 0.02 mg/100 g of protein to
 257 1.31 ± 0.11 mg/100 g of protein and 471.9 ± 22.3 mg/100 g of protein to 639.4 ± 21.1 mg/100 g of
 258 protein, respectively. The concentration of total lysine varied from 9.89 ± 0.88 to 13.12 ± 0.78 % of
 259 total protein. In low lactose milk the content of lysine was 5.21 ± 0.30 g/100 g of protein, while the
 260 concentration of CEL and furosine was $0.28 \text{ mg} \pm 0.01 \text{ mg/100 g}$ of protein and 12.32 ± 0.31
 261 mg/100 g of protein, respectively. CML was $1.28 \text{ mg} \pm 0.11 \text{ mg/100 g}$ of protein and this value was
 262 perfectly in line with the one previously obtained. Lab scale UHT milk was prepared in order to
 263 verify the effect on raw cow milk; while the lysine content was of the same order of magnitude of
 264 the low lactose milk (4.71 ± 0.22 mg/100 g of protein), the concentration of the three markers of the
 265 MR was 18.41 ± 0.93 , 1.12 ± 0.02 and 14.41 ± 1.02 mg/100 g of protein for CML, CEL and
 266 furosine respectively. The results obtained were perfectly in line with those previously obtained for
 267 the three categories of milk (Fenaille, Parisod, Visani, Populaire, Tabet, & Guy, 2006; Tareke,
 268 Forslund, Lindh, Fahlgren, & Ostman, 2013), specifically the CML in low lactose milk was similar

269 to one previously obtained by our group for LC-MS/MS analysis (Troise, et al., 2014). The
270 concentration of CML and furosine was closed to the range previously obtained: 2.2 – 30.8 and 0.8
271 – 3.7 mg/100 g of protein for furosine and CML, respectively (de Sereys, Muller, Desic, Troise,
272 Fogliano, Acharid, et al., 2014) .

273 In bakery products CML content was 43.75 ± 2.02 and 27.15 ± 0.61 mg/100 g of protein for biscuits
274 samples and bread slices, respectively, while CEL and furosine were 46.25 ± 3.01 and $10.01 \pm$
275 0.61 and 10.91 ± 0.01 and 98.55 ± 4.61 mg/100 g of protein for biscuits and bread, respectively.
276 The lysine content was almost similar in the two products: 5.01 ± 0.04 and 5.81 ± 0.04 g/100 g of
277 protein, even if the protein content was 6% and 8% for biscuits and bread. The results here reported
278 were of the same order of magnitude as the ones previously reported. Hull et al., analyzed several
279 kinds of bread and other bakery products and the concentration of CML ranged from 2.6 to 45.1
280 mg/100 g of protein for wheaten bread and potato bread, respectively (Hull, Woodside, Ames, &
281 Cuskelly, 2012). On the other hand He and coworker reported higher values for wholemeal bread:
282 CML ranged from 66.72 to 109.9 mg/100g of protein and CEL ranged from 53.30 to 82.04 mg/100
283 g protein for bread, while in biscuits samples the concentrations varied from 50.8 to 116.7 and
284 15.87 to 45.26 mg/100g protein for CML and CEL, respectively (He, Zeng, Zheng, He, & Chen,
285 2014). Interestingly, the concentration of furosine in bread (after 20 min at 200° C) is similar to the
286 one reported by Capuano and coworker: after 13 min the concentration of furosine increased up to
287 200 mg/100 g of protein and it quickly decreased up to 20 mg/100 g protein at the end of the
288 thermal treatment (Capuano, Ferrigno, Acampa, Ait-Ameur, & Fogliano, 2008). A similar kinetic
289 profile was observed also by Ramirez-Jimenez and coworker in sliced bread: the concentration of
290 furosine at the end of the process was 79.3 mg/100 g of protein while after 12 min it reached a
291 concentration higher than 200 mg/100 g protein (Ramirez-Jimenez, Garcia-Villanova, & Guerra-
292 Hernandez, 2001). In biscuit samples the kinetic profile revealed similar trends to the ones obtained
293 for bread; as a consequence at the end of the thermal process the concentration of furosine value of

294 10.01 \pm 0.61 mg/100 g of protein was comparable to those of sucrose-containing cookies reported
 295 by previous authors (Gökmen, Serpen, Açar, & Morales, 2008).

296 The above described analytical performances were evaluated in industrially prepared soybean feeds
 297 in order to verify the main advantages of the method on industrial sampling. The simultaneous
 298 quantification of the four analytes allowed a direct overview of the extent of the MR, where the
 299 concentration of lysine and the formation of furosine, CEL and CML can be easily related to the
 300 final quality of foods using a single extraction and a single injection. According to the procedure
 301 described in material and methods section, soybeans were incubated at 110° C for one hour in an
 302 autoclave and the kinetic profile was reported in **Figure 3**. The initial concentration of lysine was
 303 3.45 \pm 0.12 g/100 g of protein while CML, CEL and furosine were 9.94 \pm 0.74, 0.98 \pm 0.04 and
 304 24.24 \pm 1.74 mg/100 g of protein respectively. After 30 minutes the concentration of furosine
 305 reached the highest values: 108.01 \pm 8.97, then it rapidly decreased up to 60.58 \pm 3.75 mg/100 g of
 306 protein after 55 min. According to the reaction mechanism the degradation of the Amadori products
 307 was followed by the increase of CML: at the end of the thermal treatment its concentration was
 308 higher than 76 mg/100 g of protein. CEL reached the maximum concentration after 45 minutes
 309 (2.41 \pm 0.24 mg/100 g of protein), then it decreased probably due to degradation processes or to the
 310 blockage of methylglyoxal by other compounds. The degradation of lysine was constant throughout
 311 the thermal treatment, after 60 min lysine concentration was 2.60 \pm 0.08 g/100 g of protein thus
 312 around 23%. Several studies reported the effect of soy proteins in the development of the MR
 313 focusing on soy health benefits and on the presence of functional molecules able to control the
 314 extent of the MR (Palermo, Fiore, & Fogliano, 2012).

315 This paper represents the first example of a systematic study on the relationship between thermal
 316 treatments, MR and soybean products in feeds and in pet food a topic recently attracting the
 317 attention of the scientific community. In fact, it has been observed that the average daily intake
 318 (mg/kg body weight^{0.75}) of HMF is 122 times higher for dogs and 38 times higher for cats than

average intake for adult humans. Possible health risks, such as diabetes and renal failure, can be associated to the intake of MRPs not only in humans, but also in pets (van Rooijen, Bosch, van der Poel, Wierenga, Alexander, & Hendriks, 2013).

4. Conclusion

The analytical method allowed a comprehensive approach in the analysis of MRPs, simultaneously determining both lysine and its heat-induced derivatives. Up to now the golden standards for MRPs detection were RP-HPLC with UV-vis detection for furosine and LC-MS/MS for CML, CEL and lysine, respectively. These results showed that the extraction procedure with nitrogen and hydrochloric acid provided a good compromise for the simultaneous detection of the four analytes. The analytical performances showed high sensitivity and good reproducibility and repeatability in several foods. Quantitative data were fully in line with those previously obtained by other authors on similar foods. The simultaneous detection of the four analytes offered a sensitive tool for the kinetic modeling on neoformed contaminant reaction routes monitoring the precursor lysine, the intermediate furosine via the indirect analysis of the Amadori products and the end-products CEL and CML. The simultaneous monitoring of all compounds allowed to minimize the variability among different samples and to combine the reaction steps starting from lysine blockage, Amadori compounds formation and fragmentation, CML and CEL formation.

The research of Antonio Dario Troise at Wageningen University was partly supported by UniNA and Compagnia di San Paolo, in the frame of Programme STAR.

The authors declare no conflict of interests.

References

- Ahmed, M. U., Thorpe, S. R., & Baynes, J. W. (1986). Identification of N-Epsilon-Carboxymethyllysine as a Degradation Product of Fructoselysine in Glycated Protein. *Journal of Biological Chemistry*, 261(11), 4889-4894.
- Armbruster, D. A., Tillman, M. D., & Hubbs, L. M. (1994). Limit of Detection (Lod) Limit of Quantitation (Loq) - Comparison of the Empirical and the Statistical, Methods Exemplified with Gc-MS Assays of Abused Drugs. *Clinical Chemistry*, 40(7), 1233-1238.
- Bignardi, C., Cavazza, A., & Corradini, C. (2012). Determination of furosine in food products by capillary zone electrophoresis-tandem mass spectrometry. *Electrophoresis*, 33(15), 2382-2389.
- Brownlow, S., Cabral, J. H. M., Cooper, R., Flower, D. R., Yewdall, S. J., Polikarpov, I., North, A. C., & Sawyer, L. (1997). Bovine β -lactoglobulin at 1.8 Å resolution—still an enigmatic lipocalin. *Structure*, 5(4), 481-495.
- Capuano, E., Ferrigno, A., Acampa, I., Ait-Ameur, L., & Fogliano, V. (2008). Characterization of the Maillard reaction in bread crisps. *European Food Research and Technology*, 228(2), 311-319.
- Charissou, A., Ait-Ameur, L., & Birlouez-Aragon, I. (2007). Evaluation of a gas chromatography/mass spectrometry method for the quantification of carboxymethyllysine in food samples. *Journal of Chromatography A*, 1140(1-2), 189-194.
- de Sereys, A. L., Muller, S., Desic, S., Troise, A., Fogliano, V., Acharid, A., Lacotte, P., & Birlouez-Aragon, I. (2014). Potential of the FAST index to characterize infant formula quality. In V. Preedy (Ed.), *Handbook of dietary and nutritional aspects of bottle feeding* : Wageningen Academic Publishers.
- Delatour, T., Hegele, J., Parisod, V., Richoz, J., Maurer, S., Steven, M., & Buetler, T. (2009). Analysis of advanced glycation endproducts in dairy products by isotope dilution liquid chromatography-electrospray tandem mass spectrometry. The particular case of carboxymethyllysine. *Journal of Chromatography A*, 1216(12), 2371-2381.
- Erbersdobler, H. F., & Somoza, V. (2007). Forty years of furosine - Forty years of using Maillard reaction products as indicators of the nutritional quality of foods. *Molecular Nutrition & Food Research*, 51(4), 423-430.
- Fenaille, F., Parisod, V., Visani, P., Populaire, S., Tabet, J. C., & Guy, P. A. (2006). Modifications of milk constituents during processing: A preliminary benchmarking study. *International Dairy Journal*, 16(7), 728-739.
- Fiore, A., Troise, A. D., Mogol, B. A., Roullier, V., Gourdon, A., Jian, S. E., Hamzalioglu, B. A., Gokmen, V., & Fogliano, V. (2012). Controlling the Maillard Reaction by Reactant Encapsulation: Sodium Chloride in Cookies. *Journal of Agricultural and Food Chemistry*, 60(43), 10808-10814.
- Frolov, A., & Hoffmann, R. (2008). Separation of Amadori peptides from their unmodified analogs by ion-pairing RP-HPLC with heptafluorobutyric acid as ion-pair reagent. *Analytical and Bioanalytical Chemistry*, 392(6), 1209-1214.
- Gökmen, V., Serpen, A., Açar, Ö. Ç., & Morales, F. J. (2008). Significance of furosine as heat-induced marker in cookies. *Journal of cereal science*, 48(3), 843-847.
- Hartkopf, J., Pahlke, C., Ludemann, G., & Erbersdobler, H. F. (1994). Determination of N-Epsilon-Carboxymethyllysine by a Reversed-Phase High-Performance Liquid-Chromatography Method. *Journal of Chromatography A*, 672(1-2), 242-246.
- He, J. L., Zeng, M. M., Zheng, Z. P., He, Z. Y., & Chen, J. (2014). Simultaneous determination of N (epsilon)-(carboxymethyl) lysine and N (epsilon)-(carboxyethyl) lysine in cereal foods by LC-MS/MS. *European Food Research and Technology*, 238(3), 367-374.

- 382 Henle, T., Zehetner, G., & Klostermeyer, H. (1995). Fast and Sensitive Determination of Furosine. *Zeitschrift Fur*
383 *Lebensmittel-Untersuchung Und-Forschung*, 200(3), 235-237.
- 384 Hidalgo, F. J., & Zamora, R. (2005). Interplay between the Maillard reaction and lipid peroxidation in biochemical
385 systems. *Maillard Reaction: Chemistry at the Interface of Nutrition, Aging, and Disease*, 1043, 319-326.
- 386 Hodge, J. E. (1953). Chemistry of Browning Reactions in Model Systems. *J. Agric. Food Chem.*(1), 928-943.
- 387 Hull, G. L. J., Woodside, J. V., Ames, J. M., & Cuskelly, G. J. (2012). N-epsilon-(carboxymethyl)lysine content of
388 foods commonly consumed in a Western style diet. *Food Chemistry*, 131(1), 170-174.
- 389 Krause, R., Knoll, K., & Henle, T. (2003). Studies on the formation of furosine and pyridosine during acid hydrolysis of
390 different Amadori products of lysine. *European Food Research and Technology*, 216(4), 277-283.
- 391 Leclerc, J., Birlouez-Aragon, I., & Meli, M. (2002). Fortification of milk with iron-ascorbate promotes lysine glycation
392 and tryptophan oxidation. *Food Chemistry*, 76(4), 491-499.
- 393 Meltretter, J., Becker, C. M., & Pischetsrieder, M. (2008). Identification and site-specific relative quantification of beta-
394 lactoglobulin modifications in heated milk and dairy products. *Journal of Agricultural and Food Chemistry*,
395 56(13), 5165-5171.
- 396 Mossine, V. V., & Mawhinney, T. P. (2007). N-alpha-(1-DeOXY-D-fructos-1-yl)-L-histidine ("D-fructose-L-
397 histidine"): a potent copper chelator from tomato powder. *Journal of Agricultural and Food Chemistry*, 55(25),
398 10373-10381.
- 399 Nguyen, H. T., van der Fels-Klerx, H. J., & van Boekel, M. A. J. S. (2013). N-epsilon-(carboxymethyl)lysine: A Review on
400 Analytical Methods, Formation, and Occurrence in Processed Food, and Health Impact. *Food Reviews*
401 *International*, 30(1), 36-52.
- 402 Niquet-Leridon, C., & Tessier, F. J. (2011). Quantification of N-epsilon-carboxymethyl-lysine in selected chocolate-
403 flavoured drink mixes using high-performance liquid chromatography-linear ion trap tandem mass
404 spectrometry. *Food Chemistry*, 126(2), 655-663.
- 405 Palermo, M., Fiore, A., & Fogliano, V. (2012). Okara promoted acrylamide and carboxymethyl-lysine formation in
406 bakery products. *J Agric Food Chem*, 60(40), 10141-10146.
- 407 Pischetsrieder, M., & Henle, T. (2012). Glycation products in infant formulas: chemical, analytical and physiological
408 aspects. *Amino Acids*, 42(4), 1111-1118.
- 409 Ramirez-Jimenez, A., Garcia-Villanova, B., & Guerra-Hernandez, E. (2001). Effect of toasting time on the browning of
410 sliced bread. *Journal of the Science of Food and Agriculture*, 81(5), 513-518.
- 411 Tareke, E., Forslund, A., Lindh, C. H., Fahlgren, C., & Ostman, E. (2013). Isotope dilution ESI-LC-MS/MS for
412 quantification of free and total N epsilon-(1-Carboxymethyl)-L-Lysine and free N epsilon-(1-Carboxyethyl)-L-
413 Lysine: Comparison of total N epsilon-(1-Carboxymethyl)-L-Lysine levels measured with new method to
414 ELISA assay in gruel samples. *Food Chemistry*, 141(4), 4253-4259.
- 415 Technische Universität Dresden. (2014). AGE Database. In, vol. 2014).
- 416 Tessier, F. J., & Birlouez-Aragon, I. (2012). Health effects of dietary Maillard reaction products: the results of ICARE
417 and other studies. *Amino Acids*, 42(4), 1119-1131.
- 418 Troise, A. D., Dathan, N. A., Fiore, A., Roviello, G., Di Fiore, A., Caira, S., Cuollo, M., De Simone, G., Fogliano, V.,
419 & Monti, S. M. (2014). Faax enzymes inhibited Maillard reaction development during storage both in protein
420 glucose model system and low lactose UHT milk. *Amino Acids*, 46(2), 279-288.

- 421 Troise, A. D., Fiore, A., Colantuono, A., Kokkinidou, S., Peterson, D. G., & Fogliano, V. (2014). Effect of olive mill
422 wastewater phenol compounds on reactive carbonyl species and maillard reaction end-products in ultrahigh-
423 temperature-treated milk. *J Agric Food Chem*, 62(41), 10092-10100.
- 424 Troise, A. D., Fiore, A., Roviello, G., Monti, S. M., & Fogliano, V. (2014). Simultaneous quantification of amino acids
425 and Amadori products in foods through ion-pairing liquid chromatography-high-resolution mass spectrometry.
426 *Amino Acids*.
- 427 Vallejo-Cordoba, B., Mazorra-Manzano, M. A., & Gonzalez-Cordova, A. F. (2004). New capillary electrophoresis
428 method for the determination of furosine in dairy products. *Journal of Agricultural and Food Chemistry*,
429 52(19), 5787-5790.
- 430 van Boekel, M., Fogliano, V., Pellegrini, N., Stanton, C., Scholz, G., Lalljie, S., Somoza, V., Knorr, D., Jasti, P. R., &
431 Eisenbrand, G. (2010). A review on the beneficial aspects of food processing. *Molecular Nutrition & Food*
432 *Research*, 54(9), 1215-1247.
- 433 van Rooijen, C., Bosch, G., van der Poel, A. F. B., Wierenga, P. A., Alexander, L., & Hendriks, W. H. (2013). The
434 Maillard reaction and pet food processing: effects on nutritive value and pet health. *Nutrition research reviews*,
435 26(02), 130-148.
- 436 Vitaglione, P., Lumaga, R. B., Stanzione, A., Scalfi, L., & Fogliano, V. (2009). beta-Glucan-enriched bread reduces
437 energy intake and modifies plasma ghrelin and peptide YY concentrations in the short term. *Appetite*, 53(3),
438 338-344.
- 439 Yalcin, T., & Harrison, A. G. (1996). Ion chemistry of protonated lysine derivatives. *Journal of mass spectrometry*,
440 31(11), 1237-1243.
- 441 Yaylayan, V. A., & Huyghuesdespointes, A. (1994). Chemistry of Amadori Rearrangement Products - Analysis,
442 Synthesis, Kinetics, Reactions, and Spectroscopic Properties. *Critical Reviews in Food Science and Nutrition*,
443 34(4), 321-369.

444

445

446

447

448

449

450

451

452

453

454 **Figure legend**

455 **Figure 1:** Effect of glucose and dicarbonyls on the formation of protein-bound MRPs. At the
456 bottom the structure of β -Lactoglobulin (Brownlow, Cabral, Cooper, Flower, Yewdall, Polikarpov,
457 et al., 1997).

458 **Figure 2:** Extracted ion chromatogram of the four target molecules and their respective internal
459 standards

460 **Figure 3:** Kinetic profile of the precursor lysine (green), intermediate, furosine (blue) and end-
461 products, CML and CEL (red).

462 **Figure S1:** Fragmentation pathway for lysine and its deuterated internal standard $d4$ -lysine. The
463 structures of pipecolic acid and 1,2,3,4-tetrahydropyridin-1-ium ion was reported (Yalcin &
464 Harrison, 1996).

465

466 **Table legend**

467 **Table 1:** Mass spectrometry set up

468 **Table 2:** Analytical performances for the four analytes and their respective internal standards

469 **Table 3:** MRPs concentration after 8 replicates in different samples, the results for CML, CEL and
470 furosine were reported as mg/100 g of protein, except for lysine. The results were compared to the
471 AGE Database (Technische Universität Dresden, 2014).

Tables

Table 1

Compounds	[M+H] ⁺	Fragments	CE (V)	DP (V)
CML	205	84	29	30
		130.2	27	30
<i>d</i> ₂ -CML	207	84	30	20
		144	21	20
		130	17	20
Furosine	255.1	130	18	21
		84.4	28	21
Lys	147.2	130.2	16	30
		84.1	24	30
<i>d</i> 4-Lys	151.3	134.1	15	30
		88.2	26	30
CEL	219.2	130.3	20	30
		84.0	28	30
<i>d</i> 4-CEL	223	134.1	18	25
		88.0	30	25

Table 1: Mass spectrometry set up

Table 2

Compound	LOD	LOQ	RSD [%]	Linearity range	r ²	Recovery
CML	0.5 ppb	5 ppb	7	5-1000 ng/ml	> 0.99	91.1 ± 8.4
CEL	1 ppb	5 ppb	5	5-1000 ng/ml	> 0.99	84.2 ± 7.4
Lysine	0.5 ppb	5 ppb	5	5-1000 ng/ml	> 0.99	88.0 ± 6.9
Furosine	3 ppb	9 ppb	8	9-1000 ng/ml	> 0.99	88.0 ± 6.9

Table 2: Analytical performances for the four analytes and their respective internal standards

Table 3:

Food	CML	CEL	Furosine	Lysine (g/100 g protein)
Infant formula -1	8.22 ± 0.31	0.71 ± 0.02	471.91 ± 22.31	9.89 ± 0.88
Infant formula -2	10.4 ± 0.52	0.85 ± 0.06	542.53 ± 11.91	12.24 ± 0.91
Infant formula -3	10.9 ± 1.03	1.10 ± 0.05	574.5 ± 44.12	13.12 ± 0.78
Infant formula -4	14.81 ± 0.92	1.31 ± 0.11	639.4 ± 21.11	10.28 ± 1.01
<i>Age Database</i>	0.6 – 40.5	/	Up to 1819	/
Low lactose milk	1.28 ± 0.11	0.28 ± 0.01	12.32 ± 0.31	5.21 ± 0.30
<i>Age Database</i>	1.4	/	/	/
Lab scale UHT milk	18.41 ± 0.93	1.12 ± 0.02	14.41 ± 1.02	4.71 ± 0.22
<i>Age Database</i>	0.9-8.3	/	12.4 – 220.0	/
Biscuits	43.75 ± 2.02	46.25 ± 3.01	10.01 ± 0.61	5.01 ± 0.04
Bread slices	27.15 ± 0.61	10.91 ± 0.01	98.55 ± 4.61	5.81 ± 0.04
<i>Age Database</i>	2.6 – 45.1	/	/	/

Table 3: MRPs concentration after 8 replicates in different samples, the results for CML, CEL and furosine were reported as mg/100 g of protein, except for lysine. The results were compared to the AGE Database (Technische Universität Dresden, 2014).

Figure-1

Maillard Reaction

Acidic Hydrolysis

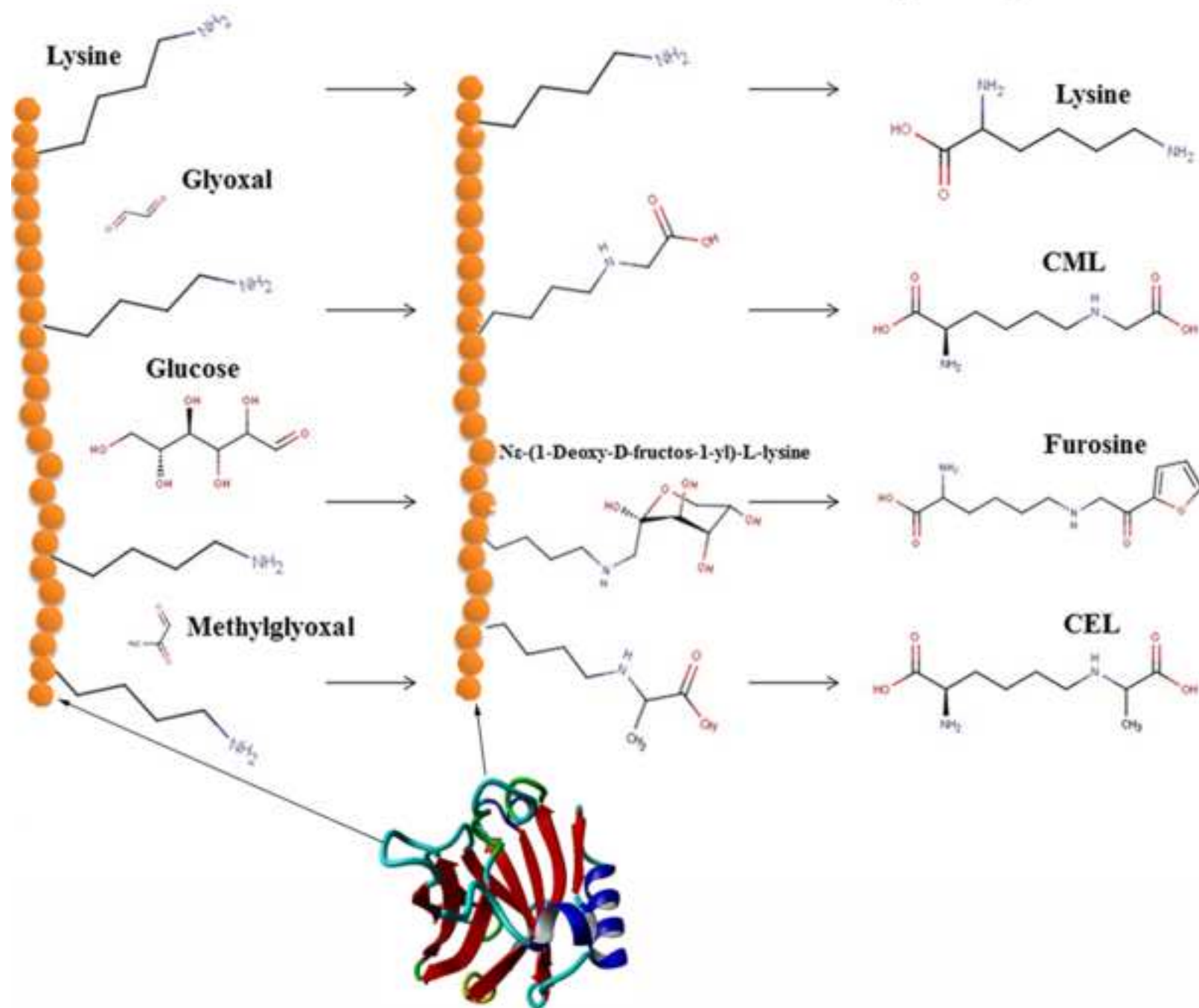


Figure-2

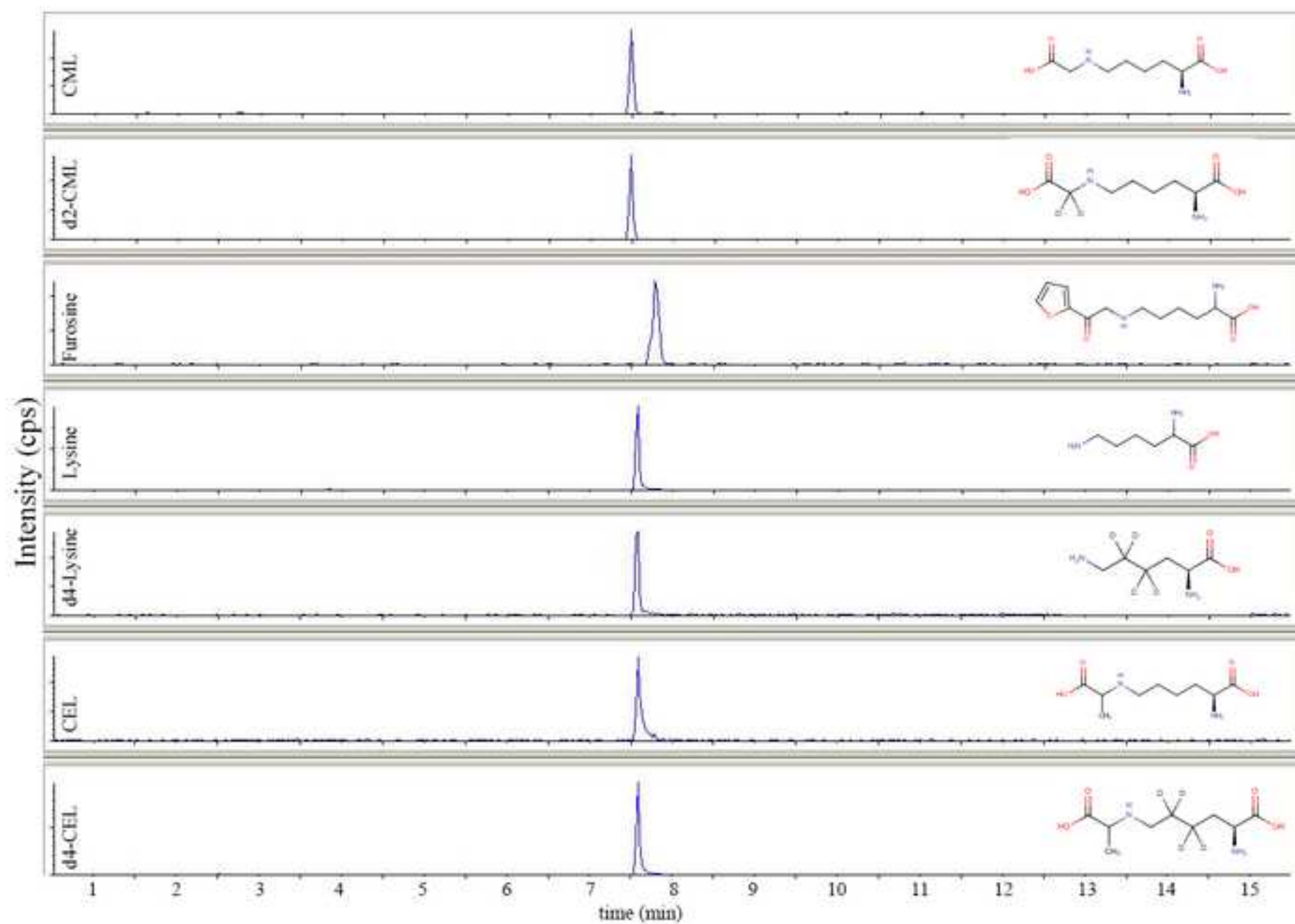
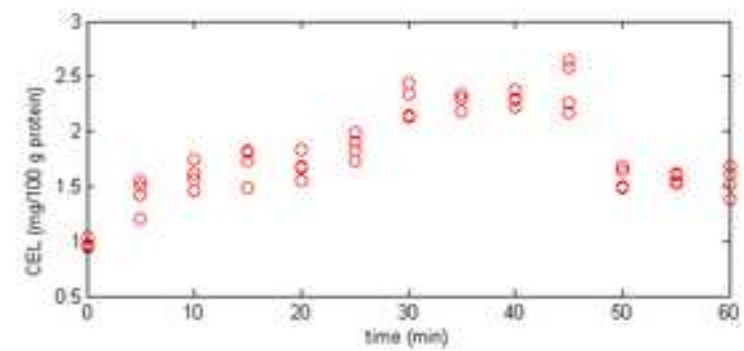
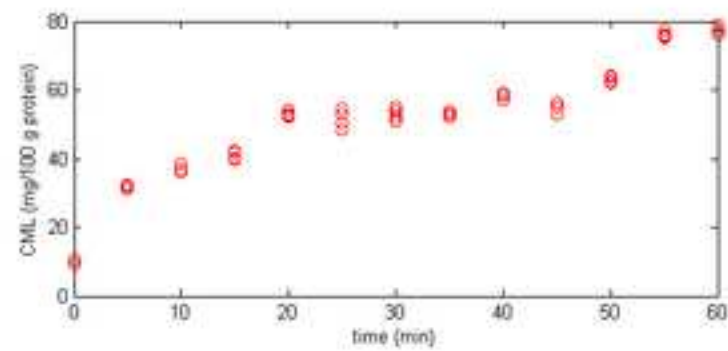
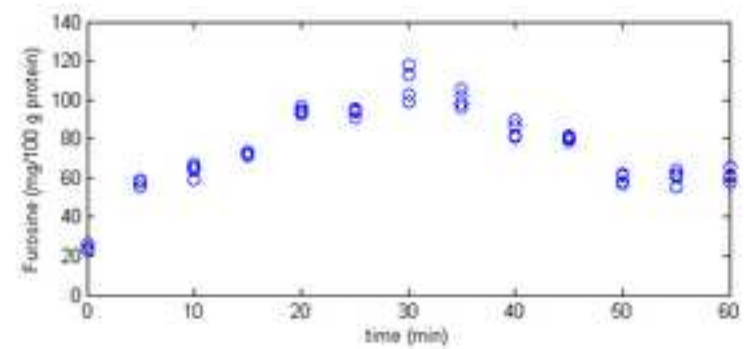
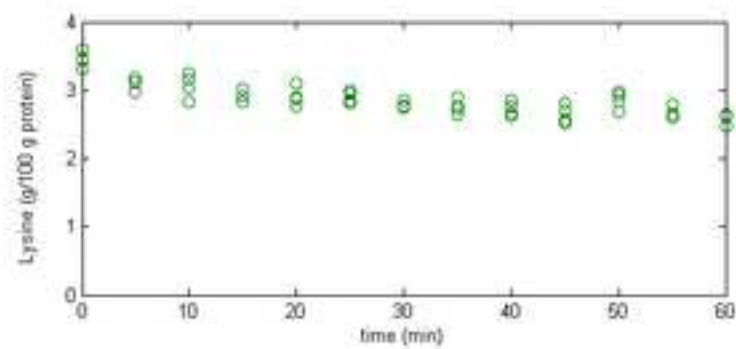


Figure-3

ACCEPTED MANUSCRIPT



- Tandem mass spectrometry and stable isotope dilution ensured reliable performances.
- The method achieved simultaneous detection of CML, CEL, Lysine and furosine.
- CML, CEL, Lysine and furosine were quantified in several foods.
- The analysis of the four markers paved the way for a better quality control.